

Notes

A New Alkaloid from Two Coccinellid Beetles *Harmonia axyridis* and *Aiolocaria hexaspilota*

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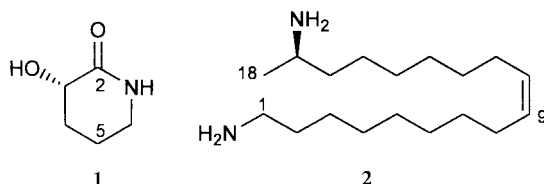
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Coccinellid beetles are known to secrete droplets of blood from the joints when they are molested.¹ A number of alkaloids are present in this reflex bleeding which is considered as a mode of their chemical defense. These alkaloids are also responsible for their aposematic coloration. About fifty alkaloids like azaphenalenones, azabicyclononanes, harmonine, pyrrolidines, piperidines, aromatic amines, azamacrolides, and some dimeric alkaloids have so far been isolated from about thirty five species of the Coccinellidae.²⁻⁴ The deterrent and toxic properties of these defensive alkaloids prompted



us to investigate *Harmonia axyridis* and *Aiolocaria hexaspilota* for their alkaloidal constituents. We here report the isolation of a new natural product 3-hydroxypiperidin-2-one (**1**) and harmonine (**2**) from these species of Coccinellidae.

Compound **1** was isolated as a light yellow solid. The $[M + H]^+$ peak was observed at m/z 116.0-712 in the HRFABMS which corresponded to the molecular formula $C_5H_9NO_2$. The formula showed two degrees of unsaturation. A broad band at 3398 cm^{-1} in the IR spectrum suggested the presence of OH and NH groups. A strong band at 1619 cm^{-1} indicated an intra-molecular hydrogen bonded carbonyl group.

The ^1H NMR of **1** displayed a proton signal at δ 3.94 that was correlated to a carbon signal at δ 62.5. This signal showed correlations with the methylene protons at δ 2.28 and 2.09. The methylene proton signals at δ 3.36 and 3.20 showed correlation with a carbon signal at δ 46.5 indicating that they are vicinal to a nitrogen function. These proton

signals showed strong correlation with the methylene proton signal at δ 1.96 that showed further correlations with the methylene proton signals at δ 2.28 and 2.09. Thus the entire proton signals in the COSY spectrum comprised a single spin system. The ^{13}C NMR spectrum featured a carbonyl carbon at δ 174.5 (C-2), an oxygenated carbon at δ 62.5 (C-3), and a carbon attached to nitrogen at δ 46.5 (C-6). The signals at δ 31.0 and 24.5 were assigned to C-4 and C-5, respectively, and ^1H - ^{13}C connectivities were confirmed by an HMQC experiment. The COSY and HMBC correlations were also in accordance with the proposed structure (Figure 1). Thus the gross structure was determined as 3-hydroxypiperidin-2-one. The compound showed an optical rotation of -53° which was the same in sign to that of the earlier synthesized (*S*)-3-hydroxypiperidin-2-one ($[\alpha]_D^{21} -6^\circ$)^{5,6} but the degree of rotation was different. The isolated compound was quite pure and the structure was well established by 2D NMR techniques so we deduce that **1** also has (*3S*)-stereochemistry. This compound has previously been synthesized⁶ but to the best of our knowledge it is the first report of its occurrence as a natural product. The reported mp of the synthetic compound varied from 133 to 171 °C, depending on its optical purity.^{5,7-9} Compound **1** has melted at 122-124 °C and decomposed at 190-192 °C.

Compound **2** was a brownish oil. Its ^1H NMR featured the presence of a terminal methyl (δ 1.04, 3H, d, $J = 7.0$ Hz, H-18), a methine (δ 2.82-2.90, 1H, m) and methylene protons (δ 2.67, 2H, t, $J = 7.0$ Hz) vicinal to a nitrogen atom, and a mono-unsaturated alkyl chain. The assignments were confirmed by COSY correlations and were supported by ^{13}C

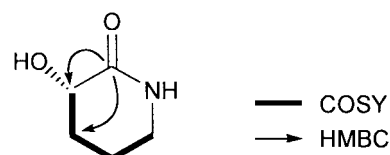


Figure 1. Key COSY and HMBC correlations for compound **1**.

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Table 1. *In Vitro* Cytotoxicities (ED₅₀ μg/mL) of **1** and **2** against Human Solid Tumor Cells^a

compound	A549	SK-OV-3	SK-MEL-2	XF498	HCT15
1	>30	>30	>30	>30	>30
cisplatin	0.72	1.23	2.26	1.03	1.10
doxorubicin	0.02	0.11	0.02	0.08	0.04
2	3.04	2.86	2.87	1.03	1.10
cisplatin	0.40	0.77	0.98	0.48	0.41
doxorubicin	0.01	0.02	0.02	0.03	0.02

^aA549: human lung cancer; SK-OV-3: human ovarian cancer; SK-MEL-2: human skin cancer; XF498: human CNS cancer; HCT15: human colon cancer. The compounds were assayed in two separate batches.

Table 2. Enzyme Inhibitory Activity of **2**^a

Concentration	Acetylcholinesterase	Prolylendopeptidase	Neuraminidase
5 μg/mL	12	53	5
40 μg/mL	47	64	12

^aExpressed as % inhibition.

NMR spectrum. The LRFABMS showed an [M + H]⁺ ion at *m/z* 283 which gave a molecular formula C₁₈H₃₈N₂ in combination with the NMR data. The location of the double bond was determined based on the significant allylic cleavages at *m/z* 168 and 114 in the FAB-CID MS/MS. These findings suggested the gross structure of compound **2** as harmonine, previously isolated from *Harmonia leis conformis* and *Hippodamia convergens*.¹⁰ The geometry of the double bond was determined as cis on the basis of the chemical shift value of the allylic carbons (see Experimental) while the absolute stereochemistry was presumed to be the same as that of harmonine.¹¹

Compounds **1** and **2** were evaluated for their cytotoxicity against five human solid tumor cell lines, and **2** was found to possess significant cytotoxicity against these cell lines (Table 1). Compound **2** was also evaluated for its inhibitory activity on several enzymes that are considered as therapeutic targets of Alzheimer's disease. It showed a weak inhibitory activity against acetylcholinesterase (AChE), prolylendopeptidase (PEP), and neuraminidase (Table 2).

Experimental Section

General Experimental Procedures. Melting point was measured on an Electrothermal digital melting point apparatus and was uncorrected. Optical rotations were measured on a JASCO DIP-370 digital polarimeter. IR spectra were obtained using a JASCO FT/IR-410 spectrometer. ¹H and ¹³C NMR spectra were recorded on a Varian Unity Plus 300, and Varian INOVA 500 spectrometers. Chemical shifts were reported in reference to the respective residual solvent peaks (δ_H 3.3 and δ_C 49.0 for CD₃OD). COSY spectra were recorded on a Varian INOVA 500 spectrometer and Bruker DMX 600 spectrometer for compound **1** and **2**, respectively. HRFABMS data were obtained using a JEOL JMS-HX110/

110A. HPLC was performed on a Gilson 370 pump with a YMC amino 12S05 (250 × 10 mm i.d., S-5 μm, 120 Å) and YMC ODS-H80 (250 × 10 mm i.d., S-4 μm, 80 Å) column using a Shodex RI-71 detector.

Animal Material. About 1000 adult ladybird beetles were collected and were kept at -20 °C until extraction.

Isolation of Compounds. The frozen beetles were extracted with MeOH at room temp. and the combined extract was filtered and vacuum dried on a rotary evaporator to give a thick paste (9 g). Reversed-phase flash chromatographic separation of this extract with 20 → 0% H₂O/MeOH and finally with acetone gave 5 fractions. Fraction 2 showed high cytotoxicity in the brine shrimp assay¹² and was further separated into 16 fractions on reversed-phase MPLC (ODS, MeOH : H₂O, 7 : 3). Fraction 2-2 was chromatographed on a reversed-phase HPLC (ODS, MeOH : H₂O, 3 : 7) into 8 fractions and repeated HPLC purification (amino, CH₃CN : H₂O, 7 : 3) of fraction 2-2-1 yielded compound **1** (94 mg). Fractions 2-4-2-16 were combined into a single fraction (fraction 2-4) on the basis of TLC and brine shrimp assay and was further separated on chromatotron (Si gel, 2 mm) eluting with a gradient of CH₂Cl₂ : MeOH : NH₃ (9 : 1 : 0.1 → 1 : 1 : 0.1) to give 7 fractions. Fraction 2-4-4 yielded compound **2** (13.5 mg) on further purification by HPLC (amino, CH₃CN : H₂O, 1 : 1).

Compound 1: Yellowish brown solid; mp 122-124 °C (decomp. 190-192 °C, lit. 171.5 °C⁵, 143-145 °C⁷, 135-137 °C⁸, 133-135 °C⁹); [α]_D¹⁸ -53° (c 0.8, MeOH); IR ν_{max} (film) 3398, 1619, 1405 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) δ 3.94 (1H, dd, *J* = 7.9, 6.4 Hz, H-3), 3.36 (1H, dt, *J* = 11.6, 5.8 Hz, H-6), 3.20 (1H, dt, *J* = 11.6, 7.3 Hz, H-6), 2.28 (1H, ddd, *J* = 12.8, 7.9, 7.9 Hz, H-4), 2.09 (1H, ddd, *J* = 12.8, 6.4, 6.4 Hz, H-4), 1.94-1.99 (2H, m, H-5); ¹³C NMR (125 MHz, CD₃OD) δ 174.5 (C-2), 62.5 (C-3), 46.5 (C-6), 31.0 (C-4), 24.5 (C-5). HRFABMS *m/z* 116.0713 [M + H]⁺ (calcd for C₅H₁₀NO₂, 116.0712, Δ +0.1 mmu)

Compound 2: Brownish oil; ¹H NMR (300 MHz, CDCl₃) δ 5.30-5.40 (2H, m, H-9, H-10), 2.82-2.90 (1H, m, H-17), 2.67 (2H, t, *J* = 7.0 Hz, H-1), 1.95 - 2.05 (4H, m, H-8, H-11), 1.11-1.40 (22H, br s, alkyl chain), 1.04 (3H, d, *J* = 7.0 Hz, H-18); ¹³C NMR (75 MHz, CDCl₃) δ 130 (C-9, C-10), 46.5 (C-17), 41.5 (C-1), 28-30 (alkyl chain), 27.0 (C-8, C-11), 23.0 (C-18); FAB-CID MS/MS *m/z* 283 [M + H]⁺ (100), 268 [M-CH₃]⁺ (3), 239 (0.1), 224 (0.1), 210 (0.1), 196 (0.1), 182 (0.2), 169 (0.2), 155 (0.1), 140 (0.1), 114 (0.2), 44 (1.2).

References

- Happ, G. M.; Eisner, T. *Science* **1961**, *134*, 329-331.
- Angela, G. K.; Meinwald, J. *Chem. Rev.* **1996**, *96*, 1105-1122.
- Lebrun, B.; Braekman, J.-C.; Daloze, D.; Kalushkov, P.; Pasteels, J. M. *Tetrahedron Lett.* **1999**, *40*, 8115-8116.
- Daloze, D.; Braekman, J.-C.; Pasteels, J. M. *Chemoecology* **1995**, *5/6*, 173-183.
- Hunter, A.; Woodward, H. E. *Biochem. J.* **1941**, *35*, 1298-1306.
- Gibbs, G.; Hateley, M. J.; McLaren, L.; Welham, M.; Willis, C. L. *Tetrahedron Lett.* **1999**, *40*, 1069-1072.
- Hua, D. H.; Zhang, F.; Chen, J.; Robinson, P. D. *J. Org. Chem.*

- 1994**, 59, 5084-5087.
8. Hjedts, H.; Honore, T. *Acta Chem. Scand.* **1978**, B32, 187-192.
9. Elming, N. *Acta Chem. Scand.* **1957**, 11, 914.
10. Braconnier, M. F.; Braekman, J.-C.; Daloze, D.; Pasteels, J. M. *Experientia* **1985**, 41, 519-520.
11. Braconnier, M. F.; Braekman, J.-C.; Daloze, D. *Bull. Soc. Chim. Belg.* **1985**, 94, 605-613.
12. Meyer, B. N.; Ferrigni, N. R.; Putnam, J. E.; Jacobsen, L. B.; Nichols, D. E.; McLaughlin, J. L. *Planta Medica* **1982**, 45, 31-34.
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